

Scalable synthesis of polymer-based atomically dispersed electrocatalysts for rechargeable zinc-air batteries

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The escalating global demand for clean energy has encouraged extensive research into advanced energy storage technologies, with rechargeable zinc-air batteries (RZABs) standing out due to their high voltage, energy density, low cost and eco-friendliness.¹ However, challenges persist regarding sluggish kinetics of the oxygen reduction reaction (ORR) and the oxygen evolution reaction (OER) at the air electrode, often mitigated with costly and scarce precious metal-based catalysts.² Transition metals offer more sustainable alternative, with recent focus on atomically dispersed catalysts (ADCs) to address issues of agglomeration and enhance efficiency.³ While promising, ADC synthesis methods and scalability pose many challenges. Polymer-derived carbons have emerged as a scalable solution, offering tailored metal environments conducive to efficient catalyst production. This study presents a novel approach to synthesise atomically dispersed bimetallic catalysts for oxygen electrocatalysis from transition metal-embedded polymer frameworks. The STEM image indicates the uniform dispersion of single-atom sites (Figure 1a). The resulting FeCoN-PDF-T₂-2 catalyst exhibits good bifunctional activity ($\Delta E = 0.75$ V) and electrochemical stability for ORR/OER showing potential for RZAB application with superior performance ($P_{\max} = 258$ mW cm⁻²) compared to PtRu/C ($P_{\max} = 185$ mW cm⁻²) (Figure 1b), thus highlighting the feasibility of large-scale production and offering new avenues for further exploration in catalyst synthesis methods.

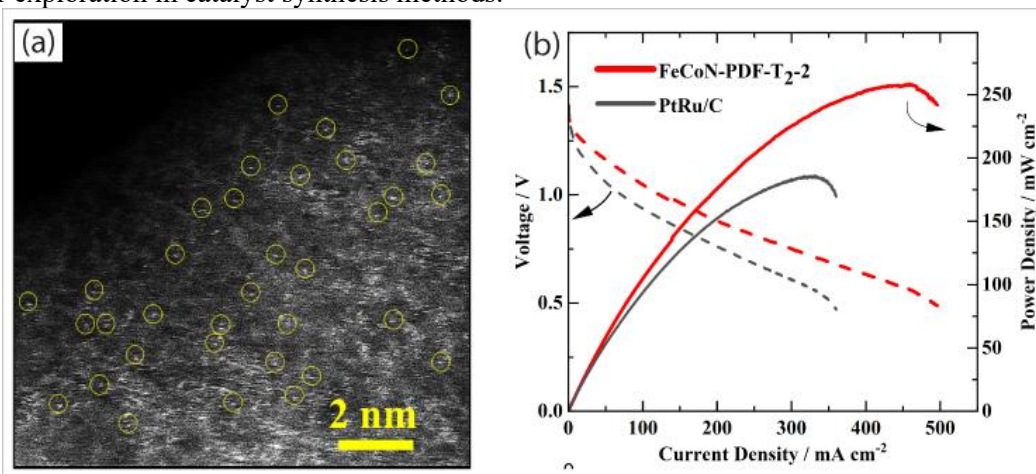


Figure 1. (a) STEM image of a FeCoN-PDF-T₂-2 catalyst, (b) discharge polarisation and power density curves of Zn-air battery with FeCoN-PDF-T₂-2 and PtRu/C air electrode.

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