

ZIF-8 based composite catalysts for oxygen reduction electrocatalysis

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Polymer electrolyte fuel cells are next-generation energy conversion devices to meet the future green-energy demands. However, the strong bond energy of O₂ molecule makes it difficult to break electrochemically which as a result leads to the slower oxygen reduction reaction (ORR) kinetics at cathode.¹ In this work, we report on Fe/Co doped and CNT supported ZIF-8 based composite catalysts for boosting the ORR in 0.1 M KOH. The catalysts (ZIF-8@CNT) were prepared via high-temperature pyrolysis at 900 °C and characterized using various physicochemical techniques and employed as cathode materials in anion exchange membrane fuel cell (AEMFC).² All the synthesized catalysts, metal-free (ZNT-900), single-metal-doped (Fe-ZNT-900, Co-ZNT-900) and binary-metal-doped (Fe₁Co₁-ZNT-900, Fe₁Co₂-ZNT-900) carry a porous morphology with a good amount of N-moieties and the presence of FeCo alloys in the carbon skeleton of bimetallic materials. The rotating disc electrode (RDE) polarization curves showed that Fe₁Co₂-ZNT-900 exhibited the maximum half-wave potential ($E_{1/2}$) of 0.85 V vs. RHE which surpassed the commercial Pt/C catalyst ($E_{1/2}$ = 0.83 V) (Figure 1a). The Koutecky–Levich plots and RRDE measurement showed that the ORR followed mainly the 4e⁻ reduction pathway with a peroxide (HO₂⁻) yield of 22.5% for Fe₁Co₂-ZNT-900 catalyst (Figure 1b). The as-synthesized catalyst materials were further tested in H₂–O₂ AEMFC, where the Fe₁Co₂-ZNT-900 cathode delivered a maximum power density (P_{max}) of 0.171 W cm⁻² and current density at 0.5 V ($j_{0.5}$) of 0.326 A cm⁻², which is very close to that of the Pt/C catalyst (P_{max} = 0.215 W cm⁻² and $j_{0.5}$ = 0.359 A cm⁻²). The prepared ZIF-8@CNT catalysts showed remarkable electrocatalytic ORR activity in 0.1 M KOH solution and fuel cell performance comparable to that of the benchmark Pt/C catalyst. All in all, the bimetallic ZIF-8@CNT catalysts performed quite well in AEMFC and the results were comparable to those of similar materials and these catalysts should be explored more in-depth to achieve more promising results in AEM fuel cells.

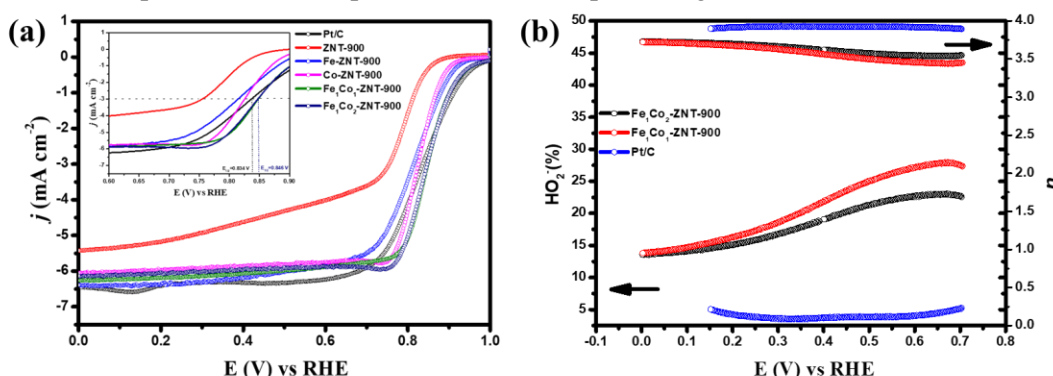


Figure 1: (a) RDE polarization curves for oxygen reduction in O₂-saturated 0.1 M KOH at 1900 rpm for all the catalyst materials and (b) HO₂⁻ yield and electron transfer number (n) for the bimetallic catalyst materials.

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References

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