

# **General aspects of mass spectrometry**

Overview of the basic theory and main features of  
mass spectrometry

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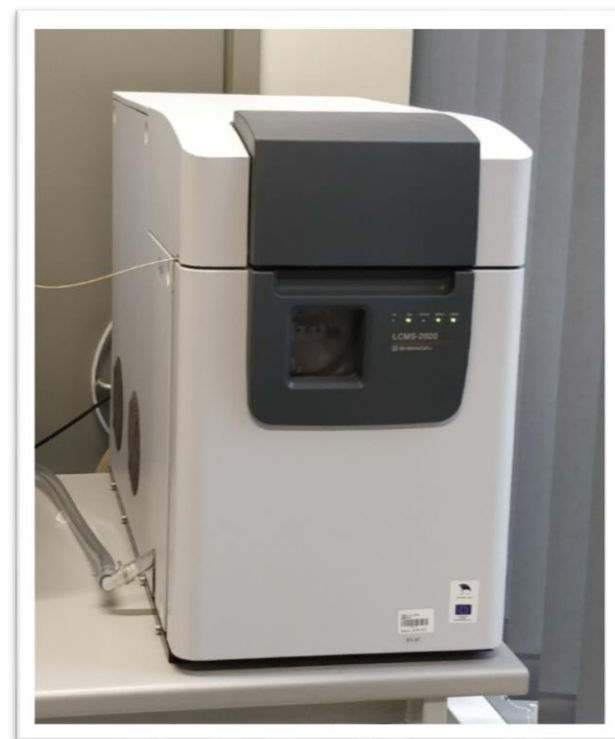
# Mass spectrometry

**Mass spectrometry** (MS) is an analytical technique for qualitative and quantitative analysis of compounds.

It is based on the separation of the components according to their **mass-to-charge ( $m/z$ ) ratio**.

## MS can be used

- for the analysis of gaseous, liquid, and solid samples.
- as an individual method (direct MS);
- as a detector for other methods (GC, LC, etc.);
- for isotope ratio analysis (GC-C-IRMS, EA-IRMS);



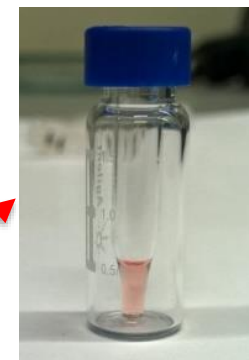
# Mass spectrometry

**MS is suitable for the analysis of:**

- pesticides
- drugs
- doping
- proteins/peptides
- lipids
- resins
- dyes
- carbohydrates/sugars
- polymers



**Horse blanket**  
Estonian National Museum  
(end of the 19th century)

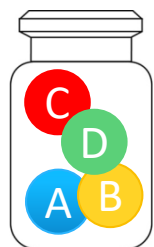


Sample solution:  
LC-MS,  
ESI/APCI/MALDI-  
FT-ICR-MS

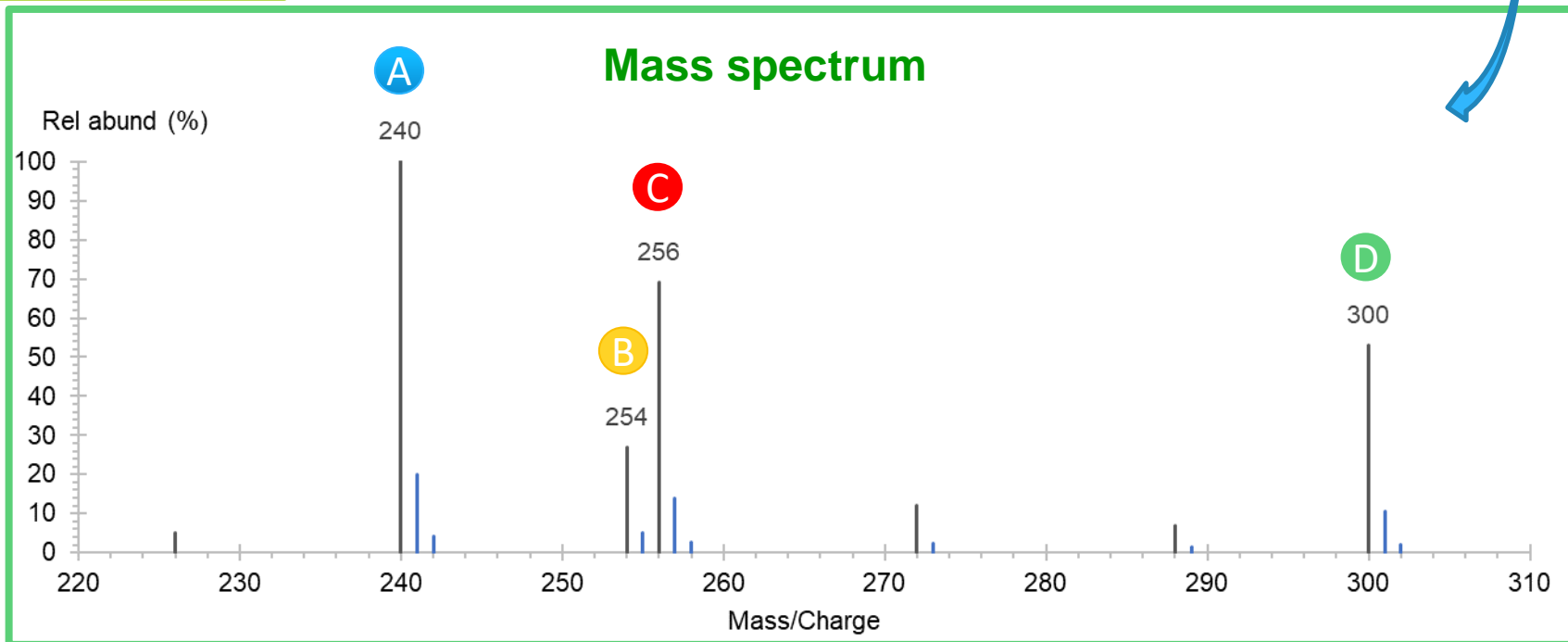
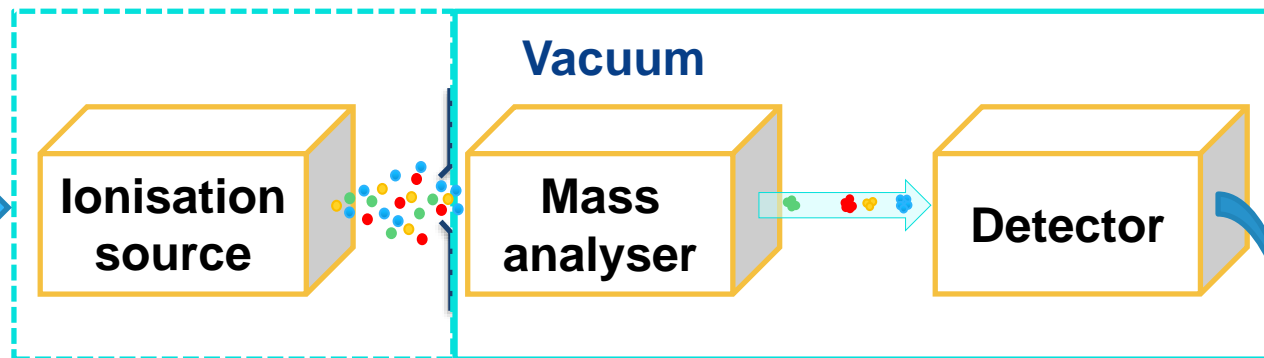


Sample for  
MALDI-FT-ICR-MS  
analysis

# Principle of mass spectrometry



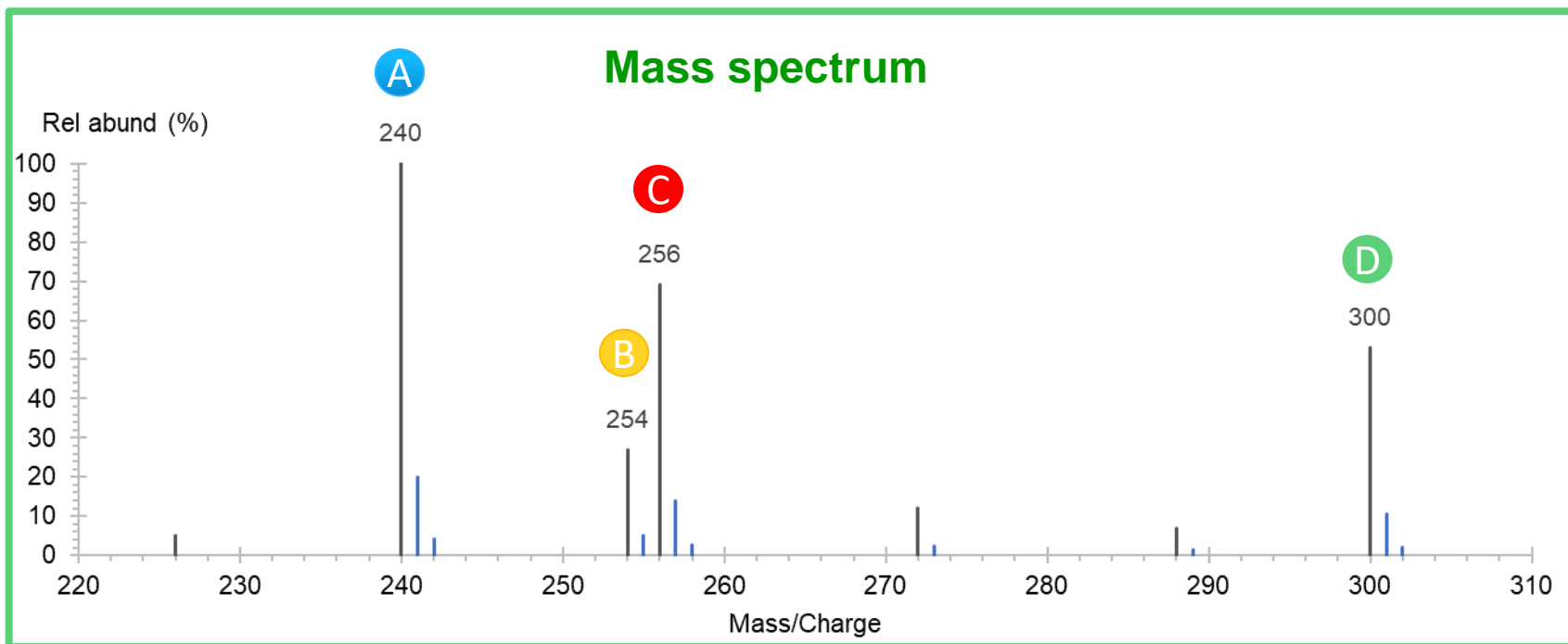
Sample (solution)  
of different  
compounds



# Information obtained with MS

## MS provides information about

- **mass-to-charge ( $m/z$ ) ratio** – **ratio of molecular mass** and **charge**, from which molecular mass of the compound can be deduced (elemental composition),
- **fragments** – information about structure,
- **isotope ratio.**



# Ionisation sources

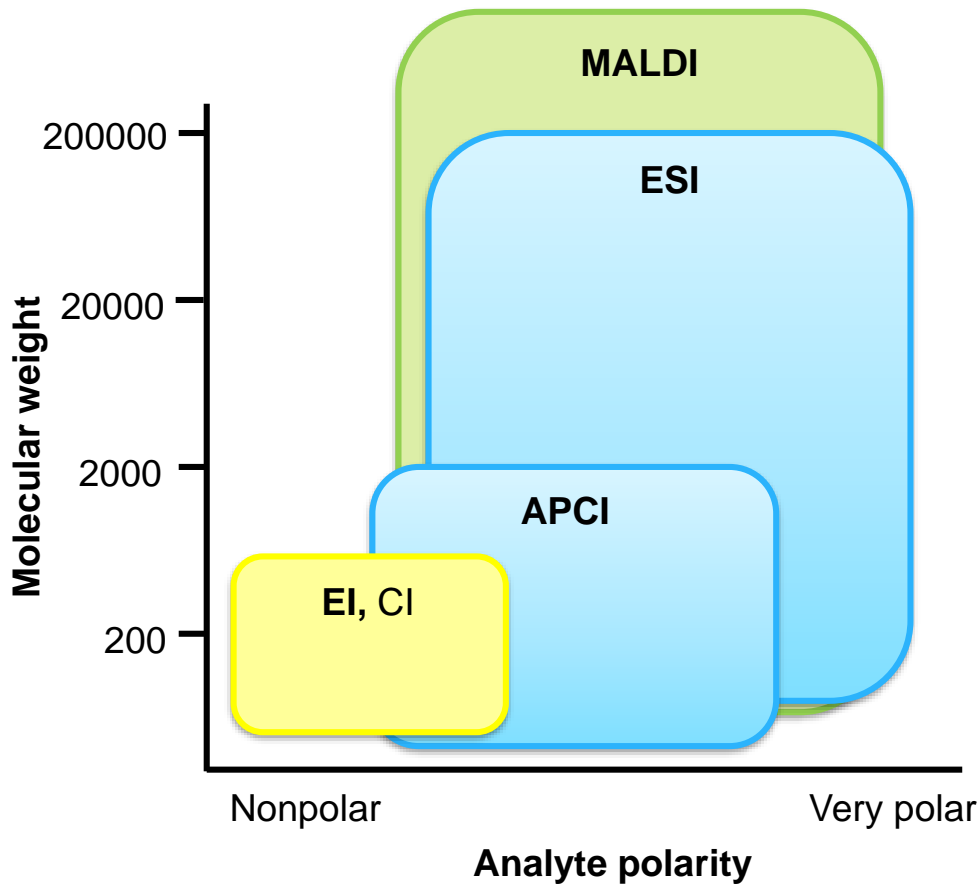
The **suitability of a ionisation source** is determined by

– **the nature of the sample:**

- the **molecular mass** of analyte,
- the **polarity** of the compounds,
- the **stability** of the sample,
- the **solubility** of the sample;

– **the type of information required**

- information about the **composition**
  - main components vs trace analysis
  - information about the contaminants
- information about the **changes in the material**
  - ageing,
  - polymerisation, etc.



\* More information about the ionisation techniques and ion formation can be found in additional materials.

# Mass analysers

MS techniques can be classified according to their ***m/z* resolution**

–shows how well the instrument can distinguish ions with similar *m/z* value

## Low Resolution MS (LRMS)

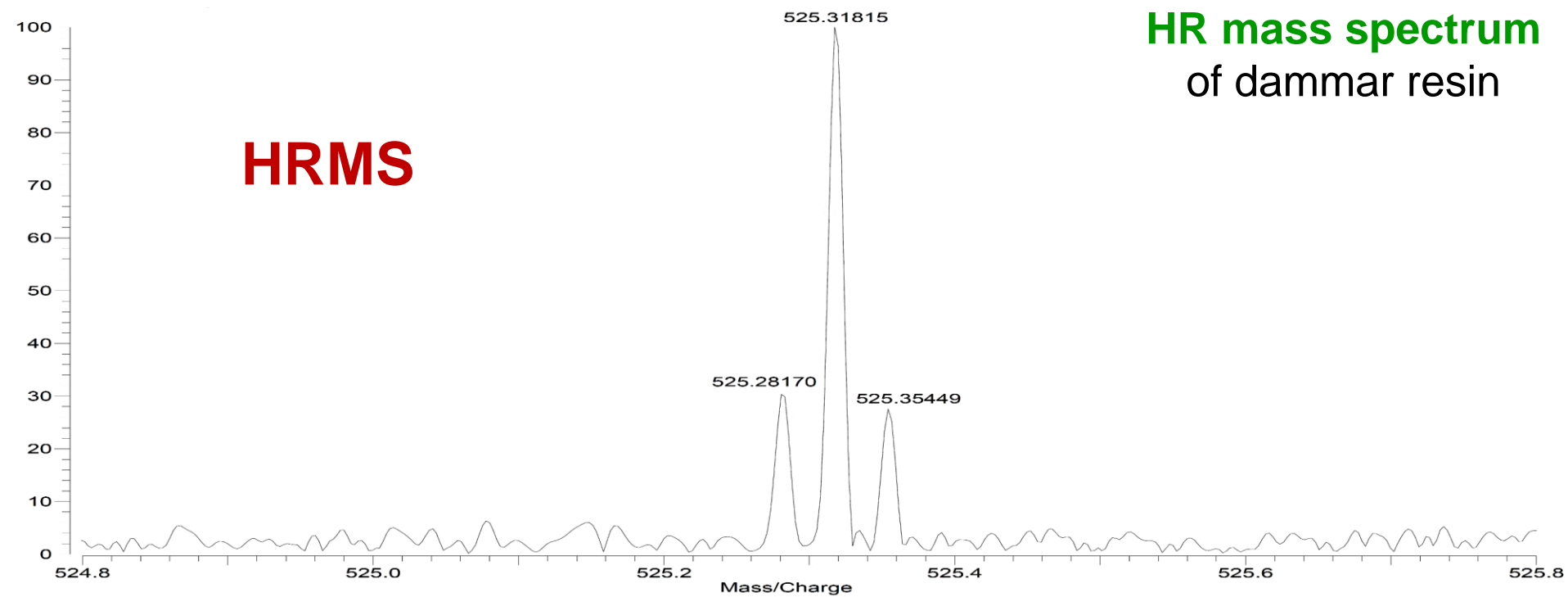
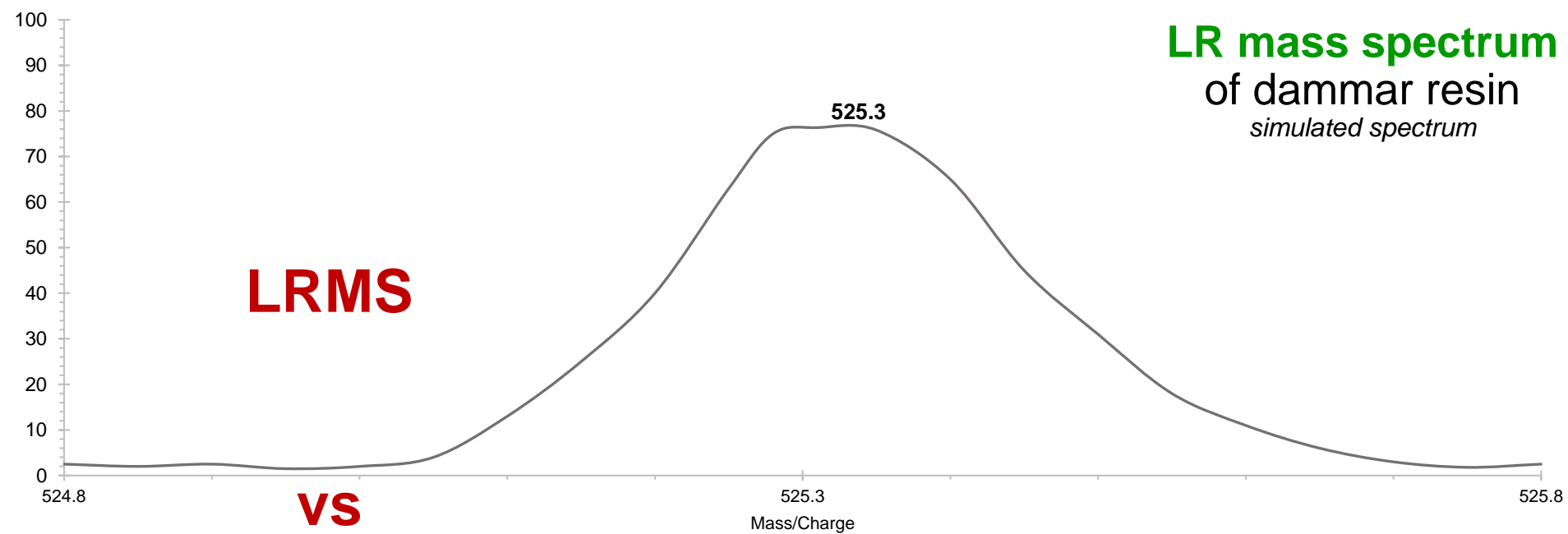
- Provides unit resolution, i.e. compounds with similar *m/z* values appear as one peak in the mass spectrum.
- **Instruments:**
  - Quadrupole (Q), incl. tandem Q
  - Ion Trap (IT)
  - Time of Flight (ToF)

## High Resolution MS (HRMS)

- Provides results with high *m/z* accuracy and therefore, distinguishes ions with very similar *m/z* values.
- **Instruments:**
  - FT-ICR-MS
  - Orbitrap
  - Time of Flight (ToF)

\* More information about different mass analysers can be found in additional materials.

Abundance Intensity (%)





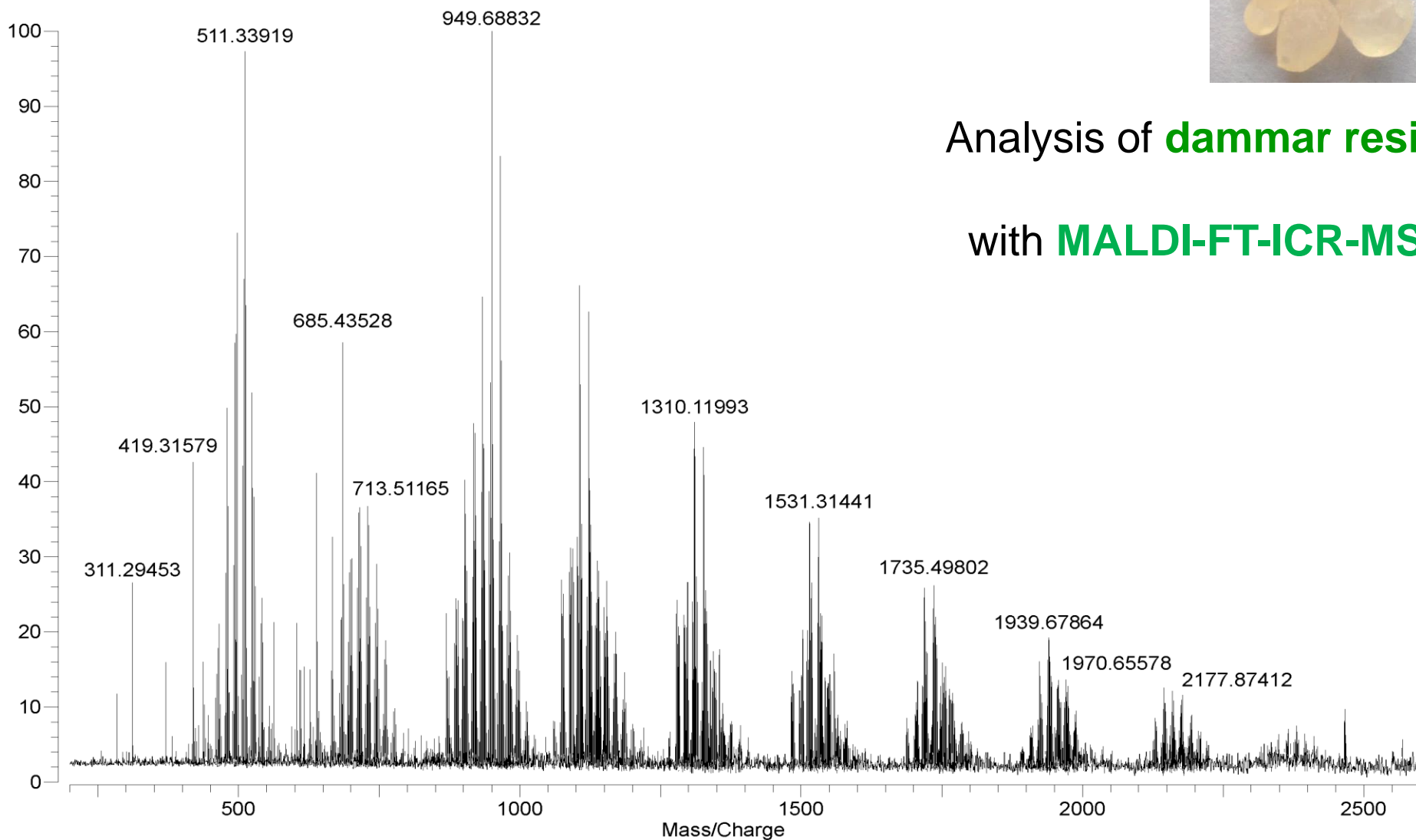
# LRMS vs HRMS



Analysis of **dammar resin**

with **MALDI-FT-ICR-MS**

Abundance Intensity (%)



Vahur, S.; Teearu, A.; Haljasorg, T.; Burk, P.; Leito, I.; Kaljurand, I. *J. Mass Spectrom.* 2012, 47(3), 392 - 409.

# Summary

**MS** – diverse and powerful group of methods suitable for the analysis of **different types of materials**.

Different **mass analysers** are compatible with various **ionisation sources**.

MS is one of the most sensitive analytical techniques – useful for **small samples** and **trace analysis**.

MS provides information about **molecular mass**, **structure**, and **isotope ratios**.