



More than just a laboratory course in advanced organic chemistry — preparation for writing a BSc thesis in style

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Dr Bartosz Trzewik

I'm a senior lecturer in the Department of Organic Chemistry, Faculty of Chemistry, Jagiellonian University in Kraków.

I have been teaching organic chemistry for nearly 20 years. Among many others I have lead:

Basic laboratory courses in organic chemistry Advanced laboratory courses in organic chemistry

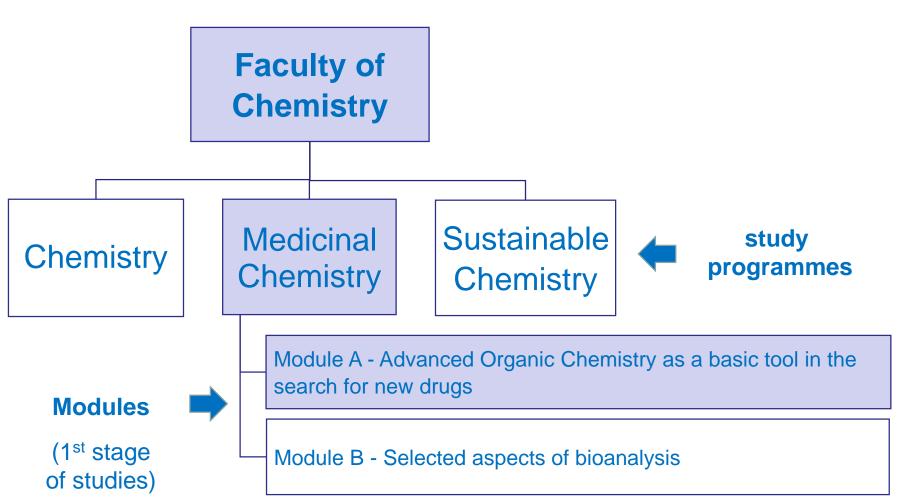
for students of study programmes: Chemistry and Medicinal Chemistry



Background

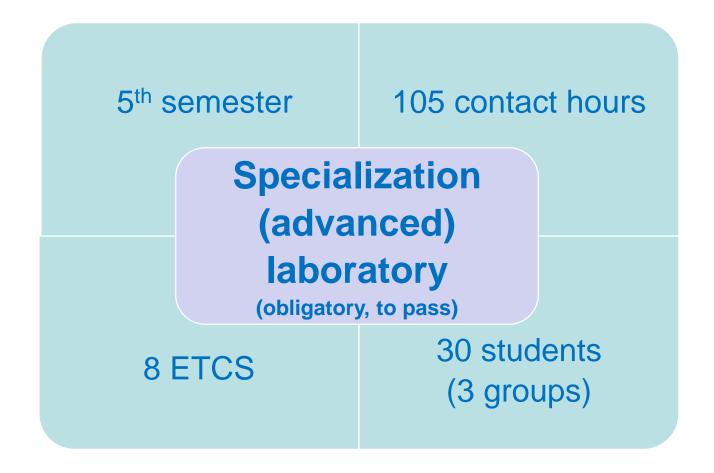


Chemistry studies at the Faculty of Chemistry





Specialization laboratory (in organic chemistry)





How to pass the course?

Preparation for the experiments (40 hours of own work)

Performing all **the experiments** (8 from 19 in each set) (105 contact hours)

Preparing reports for the experiments (50 hours of own work)

Preparing (and presenting in front of the group) **a presentation** depicting one of the experiment and its results

4th

Organic chemistry – Laboratory (basic)

semester

5th

Specialization laboratory

semester

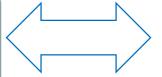
BSc thesis

6th

semester

(expected to be written in style of a scientific article – a template is provided)

Good results
Good style



Student's goal Teacher's goal



Learning Objectives

Students are expected to write their BSc thesis in style of a scientific article

BUT the students are not prepared to use the scientific style of writing

Challenge!

style of writing while preparing their reports

The students use **the template** for their BSc thesis but write their thesis in **common language and style**

Students use **the template** for their BSc thesis and write their thesis in **scientific** language

Students learn how to use the scientific

Worse results: lower quality of the text

Better results: more positive reception of the text

Routine 1



Flipped classroom approach

On site (student laboratory) or studying room

Synthesis

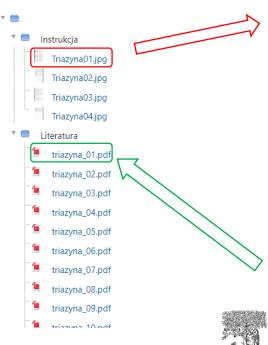
Presentation

for the group + discussion

- Instruction (in English)
- to be translated to Polish
- Literature (in English)
- to be translated to Polish
- to be studied

- Literature (in English)
- -to be studied again
- Reports
- results to be compared with other student

On-line (student educational platform or MS Teams)



On-line (student educational platform or MS Teams)



Equipment and experimental techniques involved Vial, syringe, spatula, round bottom flask, Büchner

1.3.5-Triazines or s-triazines derivatives are an important class of organic compounds that have applications in different fields such as in textile plastic and rubber industrie and are used as precided to the control of the plastic and t

finnel, filter paper, domestic kitchen-type microwave instrument, melting point apparatus, NMR spectrometer

Keywords Hexamethylmelamine [HMM], crystallization/filtration, melting point, ¹H and ¹³C NMR

spectroscopy, nucleophilic substitution, microwaves

Background

Help needed!

Literature

Available online at www.sciencedirect.com



Tetrahedron 62 (2006) 9507-9522

Tetrahedron

Tetrahedron report number 770

Recent applications of 2,4,6-trichloro-1,3,5-triazine and its derivatives in organic synthesis

Grzegorz Blotny*

Department of Chemistry and Biochemistry, University of Maryland Baltimore County, 1000 Hilltop Circle, Baltimore, MD 21250, USA

> Received 21 June 2006 Available online 14 August 2006

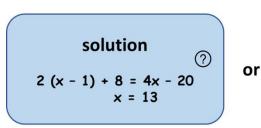
Communication – the key issue



Both should be clear

Possible sources of problems:

- Words with double / multiple meanings, different in everyday language and in scientific language.
- Limitation of a meaning:
 - salt # NaCl,
 - sugar ≠ saccharose,
 - alcohol ≠ ethanol





- Synonyms used interchangeably without explanation
- Different notations used on different items for the same concept, e.g. density quoted as ρ or d.

Possible sources of problems (continued):

- Similar words or expressions or words with a similar notation but different meaning, e.g.: chromatograph and chromatogram, limit and limitation, filter and filter off
- Improper use of italics, e.g.: m (meaning meter) instead
 of m (mass) or g (gram) instead of g (standard gravity) and
 vice versa.
- Improper use of capital letters, e.g.: m (mass) instead of M (molar mass)
- Wrong symbols, e.g.: m instead of mol

Wrong assumption that students know:

 Specialized vocabulary e.g. Grignard reaction, aliquot, defoliant

ACRONYMS, e.g. DMSO, FID, PCR, GC-MS

• EXPRESSIONS from other languages, especially Latin, e.g. vice versa, notabene, verte, implicite, explicite, vide infra, et al., op. cit., sine quanon, et caetera

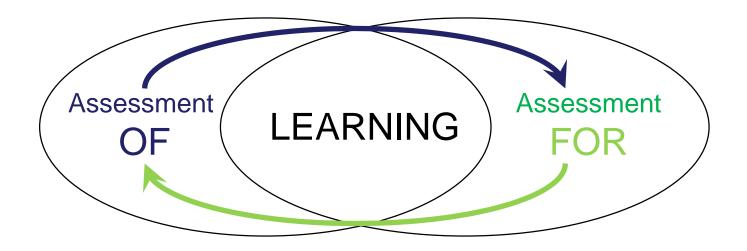


- Lack of accuracy may cause problems: some, something, somewhere, this, that means nothing
- More words does not mean better explanation
- Incorrect communication channel:
 - spoken / written word is not effective when describing the apparatus, a drawing is needed
 - drawing is not very effective when a process is shown, animation will be better,
 - spatial isomers of chemical molecules may be shown in 3D



Formative assessment as a driving force for learning





Formative assessment can precede summative assessment and should also follow it.

at lower level, each element of a student's work could be both given a mark and given a feedback (more than or instead of a mark / a grade) - for future learning



Feedback









Prompt Informative

Specific

Dialogic Complex

Comprehensive

Polite Encouraging

Delayed

Unclear

Ex Generic

cathedra Ad

personam

personam

offensive

Prejudiced



How does it work?



Concrete experience

(preparing the n-th report AND being given feedback)

Active experimentation

(preparing the (n+1)th report based on the feedback) **Reflective observation**

(reflection on the feedback)

Abstract conceptualisation

(understanding the feedback)

the same feedback must be repeated

Concrete experience

Preparing the n-th report AND being given feedback

feedback postponed

Active experimentation

(preparing the (n+1)th report based on the feedback)

(n+1)-th report after a break



Abstract conceptualisation

(understanding the feedback)

Reflective observation

(reflection on the feedback)



8

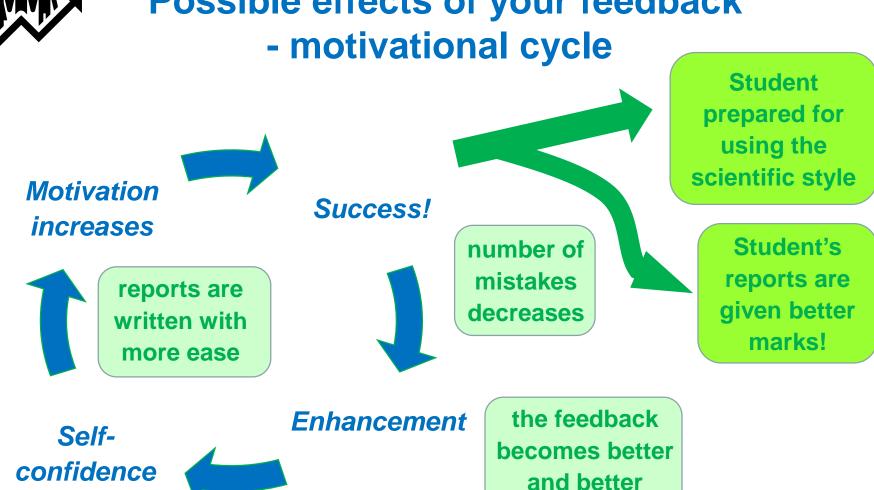
not studying the feedback



Motivational cycle



Possible effects of your feedback





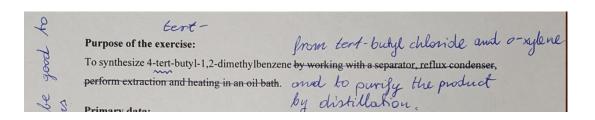
and better



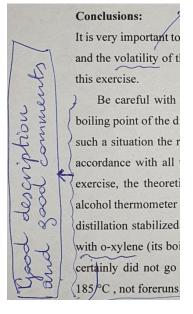
Practice (some real examples)

Content

- Informative: specific answers hints, proposals, and suggestions
- Complex: complete, touching all aspects both positive and negative



Do not just underline the mistakes!



washed with a saturated solution of sodium bicarbonate in a separating funnel, and then with a solution of sodium chloride. The organic layer was dried with anhydr. calcium chloride or magnesium sulfate. Then the drying agent was separated, the petroleum ether was distilled off on the evaporator and the remaining liquid was distilled (theoretically the distillation temperature was supposed to be 209 °C, but the measurement was done using an alcohol thermometer, which can significantly reduce the results, so the temperature during the distillation stabilized at 185 °C) under atmospheric pressure.

This is a fragment of the instruction. What drying agent did you use 2

It is befter to give the quantities of reagents immediately in cm³, grams (you can weight them) and in moles (to show molar ratios — you know which reagents are in exerts for example

(translated from Polish)

Purpose of the exercise:

To synthesize 4-tert-butyl-1,2-dimethylbenzene by working with a separator, reflux condenser,

perform extraction and heating in an oil bath. and to purify the product

by distillation.

Conclusions:

It is very important to and the volatility of the this exercise.

Be careful with boiling point of the d such a situation the r accordance with all exercise, the theoreti alcohol thermometer distillation stabilized with o-xylene (its boi certainly did not go 185°C, not foreruns.

mixture was unuted with about 20 cm³ of petroleum ether, washed with a saturated solution of sodium bicarbonate in a separating funnel, and then with a solution of sodium chloride. The organic layer was dried with anhydr. calcium chloride or magnesium sulfate. Then the drying agent was separated, the petroleum ether was distilled off on the evaporator and the remaining liquid was distilled (theoretically the distillation temperature was supposed to be 209 °C, but the measurement was done using an alcohol thermometer, which can significantly reduce the results, so the temperature during the distillation stabilized at 185 °C) under atmospheric pressure. This is a fragment of the instruction. What drying agent did you use ? It is better to give the quantities of reagents immediately in cm 3, grams (you can weight them) and in moles (to show molar ratios -> you know which reagents are in excess, for example You've chosen the glassware correctly, though the elements of the system are assigned together wrongly. Have a look at the diagram, please, and try again.

It would be good if you included a reaction diagram made in a chemical formula editor. You will be ask to do it while writing your BSc thesis

You've made a great progress! Remember about citing your references properly.

The report of execution of the task should describe the activities performed, not fragments of the instructions. Please include the observations.

Final feedback (example 2):

[Original report was written in Polish] You wrote your report almost very well, Mrs. [name]. You wrote the description of the experiment nicely. It is clearly visible that it is based on the instruction, which was translated from English with an automatic translator – please remember to check such translations. Both diagrams require links in the appropriate place in the description, as well as literature data (you must insert references in the text (eg. [1], [2], [3] ...) to articles (NOT to databases or teaching materials) and include links to these articles at the end: [1] ..., [2] ..., [3] ... The rest of the minor comments, corrections, and tips are included in the text.

Final feedback (example 1):

Mrs. [name], you have written this report quite satisfactorily, even though it is flawed:

- it would be also good to record the *observations* during the reaction and during the working-up of the reaction mixture – then it would be clear when and what gas was evolved, right?
- in the part concerning the calculation of the yield of the product, data on densities and molar masses (or masses and molar masses) of the reactants are missing. Such incomplete calculations are hard to come by;
- there is no reaction scheme.

On the other hand, you described the distillation process very well.

Summing up, I can give you no more than "B" for this report. I hope that my remarks will allow you to write the next reports better ©

What to take into account (examples)?

(during giving the feedback on the report)

- ✓ GOALS should be described specifically and completely
- ✓ REACTION SCHEME and MECHANISM drawings should apply to the UIPAC rules
- ✓ INTRODUCTION should be brief and concise, references should be included
- ✓ DEPICTION of the experiment should be complete and allow to repeat the experiment
- ✓ CALCULATIONS should be meritorically and formally correct
- ✓ CONCLUSIONS should not repeat goals and introduction



Reception

(from talks with the students and evaluation questionnaires)

Very valuable feedback!

Nobody has given me so complex feedback before.

Now I know how to show stereochemistry in chemical structures.

Dr Trzewik gave us many useful hints during the course. We haven't been taught how to cite references. Now it is clearer for me.



Summary

Advantages

- ✓ students' reports style become much better
- ✓ your feedback could be used by the students during preparing their BSc thesis
- ✓ could be used during various courses (not only organic chemistry!)
- ✓ your feedback could be used by the students during other laboratory courses
- ✓ very positive reception by the students

Disadvantages

- ✓ not suitable for larger groups
- the assistant cannot be changed during the course
- ✓ time consuming (at first)
- ✓ some elements of the feedback must be repeated (even several times)



Thank you for your attention! (and I encourage discussion)